Note

¹H NMR spectroscopic studies of the moenomycins

Lothar Hennig,¹ Matthias Findeisen,¹ Peter Welzel¹* and Rainer Haessner²

¹ Fakultät für Chemie und Mineralogie, Universität Leipzig, Talstr. 35, D-04103 Leipzig, Germany

Received 5 November 1997; revised 19 February 1998; accepted 25 February 1998

ABSTRACT: Highly resolved 1H NMR spectra of the moenomycin antibiotics can be obtained in D_2O solution when the concentration is below the critical micelle concentration. The 600 MHz spectra are structurally highly informative. \bigcirc 1998 John Wiley & Sons, Ltd.

KEYWORDS: NMR; ¹H NMR; moenomycin A, A₁₂, C₁, C₃ and C₄; aggregates; critical micelle concentration

INTRODUCTION

The moenomycins (Fig. 1) are a group of antibiotics with a unique mechanism of action. They have been shown to interfere with the enzyme(s) that catalyse the transglycosylation step of bacterial peptidoglycan biosynthesis. Most probably, their activity is related to their close structural similarities with the substrate(s) of the enzyme, and it is assumed (although not proven) that they are competitive inhibitors. These antibiotics

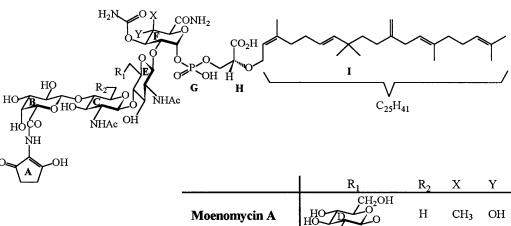
interest in the light of the growing resistance against classical antibiotics.²

Lintil recently, structural work on moreomycin A

define a new target for antibacterials, an area of great

Until recently, structural work on moenomycin A and related antibiotics was based mainly on mass and ¹³C NMR spectroscopy, ³ whereas it was impossible to obtain good quality ¹H NMR spectra. In view of ongoing work on the mode of action of these antibiotics, this was a real drawback since it would be highly desirable to obtain information on the solution conformation. Slowly the physicochemical properties of the moenomycins became better understood⁴ and it appears that the ¹H NMR problems, which are mainly associated with the amphiphilic nature of the moenomycins, can be solved. Recently, for the first time, we were able to obtain well resolved ¹H NMR spectra of a

* Correspondence to: P. Welzel, Fakultät für Chemie und Mineralogie, Universität Leipzig, Talstr. 35, D-04103 Leipzig, Germany. Contract/grant sponsor: Deutsche Forschungsgemeinschaft. Contract/grant sponsor: Fonds der Chemischen Industrie. Contract/grant sponsor: Hoechst Marion Roussel.



	Ν1	κ_2	Λ	_ I
Moenomycin A	HO TO O	Н	СН3	ОН
Moenomycin A 12	OH CH ₂ OH HO D O OH	Н	ОН	Н
Moenomycin C ₁	Н	Н	ОН	Н
Moenomycin C 3	Н	Н	CH ₃	ОН
Moenomycin C 4	ОН	Н	CH ₃	ОН

Figure 1. Structures of the moenomycins.

² Institut für Organische Chemie und Biochemie, Lehrstuhl II der TU München, Lichtenbergstr. 4, D-85748 Garching, Germany

616 HENNIG ET AL.

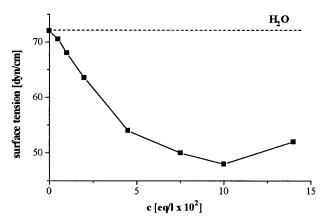


Figure 2. Surface tension of moenomycin (adapted from Ref. 6).

microemulsion consisting of moenomycin A, D_2O and C_6D_6 . Also, when moenomycin A was incorporated into SDS- d_{25} micelles, good ¹H NMR spectra could be measured. These observations allowed us to assign the resonances of nearly all non-exchangeable protons of moenomycin A (with the exception of 3-H^E, 4-H^E and 5-H^E) by means of TOCSY, HMQC, HMBC and COSY 45.⁵ However, the microemulsion method lacks generality and the SDS- d_{25} method is too expensive for everyday work.

Many years ago, the surface tension of aqueous solutions as a function of added moenomycin was determined, approaching a value of 48 dyn cm $^{-1}$ at 0.1% of moenomycin in water (Fig. 2). Although the conditions of this experiment are not very well defined, one may calculate the critical micelle concentration (cmc) of moenomycin to be in the region of 5×10^{-4} mol 1^{-1} . Work has been performed to describe the aggregation behaviour of moenomycin in water in more detail. We reasoned that at concentrations below the cmc it should be possible to obtain 1H NMR spectra of moenomycins of good quality. These expectations were confirmed by experiment, and the results obtained in these investigations are the subject of the present paper.

RESULTS AND DISCUSSION

Figure 3 shows the 200 MHz 1 H NMR spectra of moenomycin A at three different concentrations in D_2O , 1×10^{-3} , 7.5×10^{-4} and 5×10^{-4} mol 1^{-1} , i.e. in the concentration range of the cmc as discussed above. Whereas from the 1×10^{-3} mol 1^{-1} solution only the usual broad and structurally non-informative spectrum was obtained, at 7.5×10^{-4} mol 1^{-1} and even more at 5×10^{-4} mol 1^{-1} spectra of dramatically improved quality could be measured. Although the samples were lyophilized several times from D_2O solutions prior to the NMR measurements, water suppression at these very low concentrations was important and was performed by decoupler presaturation.

Figure 4 shows a 600 MHz spectrum of moenomycin A (5×10^{-4} mol 1^{-1} in D₂O) under optimized condi-

tions, demonstrating the great versatility of this approach. Similar results were obtained for all other moenomycins. Based on the previous work, all resonances of moenomycin A and the other moenomycins could be assigned and the results are given in Table 1. Since details of the assignment procedure for moenomycin A have already been described,⁵ in the current investigation the spectra of moenomycin A₁₂, C₁, C₃ and C₄ were assigned solely from 1D data, making use of the earlier assignments and restricting the discussion to the differences between the spectra of the various moenomycins.

Moenomycin A₁₂

Moenomycin A_{12} differs structurally from moenomycin A in ring F. The 4-C-methyl group is absent and ring F has a D-galacto rather than a D-gluco configuration. In the ¹H NMR spectrum, the methyl signal at $\delta=1.01$ is absent and the signals of 3-H^F and 5-H^F are clearly different from those found in moenomycin A owing to the coupling with 4-H^F: Whereas 3-H^F in moenomycin A gives a doublet, the corresponding signal in moenomycin A_{12} is a doublet of doublets. The 5-H^F signal of moenomycin A_{12} is a doublet rather than a singlet as in moenomycin A. The coupling constants $J_{3,4}=3.1$ Hz and $J_{4,5}=2.6$ Hz prove the D-galacto configuration of unit F.

Moenomycin C₁

Moenomycin C_1 belongs to the moenomycin A_{12} series (D-galacturonamide-derived unit F). Unit D is absent and unit E is derived from D-quinovosamine rather than from D-glucosamine. This leads to a methyl doublet at $\delta=1.08$, and the 5-H^E and 4-H^E signals ($\delta=3.32$ and 3.12, respectively, assigned by H,H COSY at 400 MHz) appear at higher field than in moenomycin A_{12} . The spectral data demonstrate for unit F the same substitution pattern and configuration as in moenomycin A_{12} .

Moenomycin C₃

Between moenomycin C_3 and moenomycin A there exists the same structural relationship as between moenomycin C_1 and moenomycin A_{12} , i.e. unit D is missing and unit E is derived from D-quinovosamine. This leads to the spectral features for unit E as discussed for moenomycin C_1 . The signal of 5-H^F is, as expected, a singlet at $\delta = 4.24$.

Moenomycin C₄

In this antibiotic, unit E is derived from D-glucosamine. Moenomycin C₄ is, therefore, the de-glucose derivative

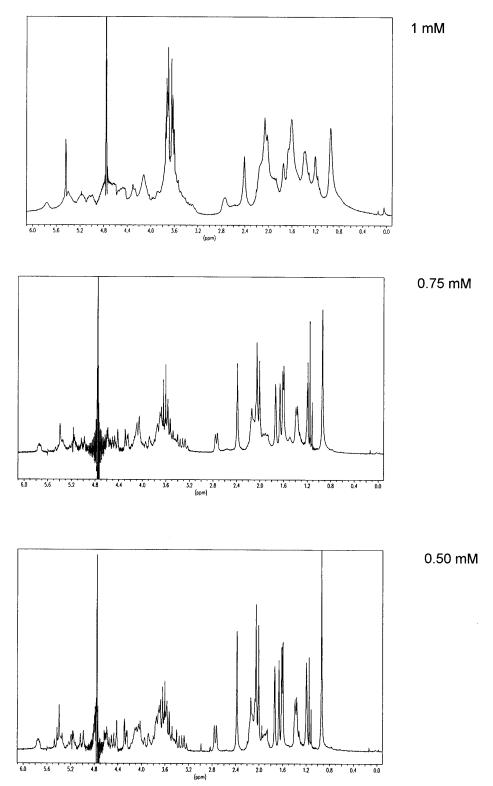


Figure 3. 200 MHz 1 H NMR spectra of moenomycin A at concentrations of 1 \times 10 $^{-3}$, 7.5 \times 10 $^{-4}$ and 5 \times 10 $^{-4}$ mol I $^{-1}$ in D $_2$ O.

of moenomycine A. The spectrum displays the expected features as discussed above (see Table 1).

Work is in progress to make use of these results in approaching the problem of the solution conformation of these antibiotics.

CONCLUSION

A method is now at hand which enables one to obtain high-quality ¹H NMR spectra of the moenomycins.

EXPERIMENTAL

¹H NMR spectra were recorded on a Gemini 200 (Varian, 199.98 MHz, processed by Bruker WINNMR

618 HENNIG ET AL.

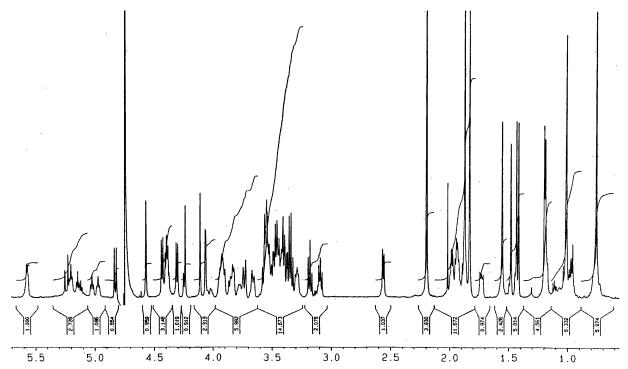


Figure 4. 600 MHz 1 H NMR spectrum of moenomycin A (5 \times 10 $^{-4}$ mol I $^{-1}$ in D2O at 285 K).

Table 1. ^{1}H NMR spectra of moenomycin A, A₁₂, C₁, C₃ and C₄

		Moenomycin						
Assignment	A	A ₁₂	C ₁	C ₃	C ₄			
CH ₃ -23 ^I ,24 ^I	0.76 s							
4 ^F -CH ₃	1.01 s	_	_	1.01 s	1.01 s			
CH_3-6^E	_	_	1.08 d	1.07 d	_			
			$(J_{5, 6} = 6.1 \text{ Hz})$	$(J_{5, 6} = 6.1 \text{ Hz})$				
CH ₂ -9 ^I	1.18 m							
CH_3-6^C	1.19 d							
	$(J_{5, 6} = 6.0 \text{ Hz})$	$(J_{5, 6} = 5.8 \text{ Hz})$	$(J_{5, 6} = 5.8 \text{ Hz})$	$(J_{5, 6} = 5.9 \text{ Hz})$	$(J_{5, 6} = 6.1 \text{ Hz})$			
CH_3-20^I	1.42 s							
CH_3-21^I	1.43 s							
CH ₃ -19 ^I	1.48 s							
CH_3-25^I	1.56 s	1.55 s	1.55 s	1.55 s	1.55 s			
CH_2-10^I	1.73 m							
CH₃CONH ^C	1.83 s	1.81 s	1.82 s	1.83 s	1.83 s			
CH ₃ CONH ^E	1.87 s	1.87 s	1.83 s	1.83 s	1.85 s			
CH_2 -15 ^I	1.87 m							
CH_2 -5 ^I ,	1.94 m	1.95 m	1.95 m	1.94 m	1.94 m			
CH_2 - 16^I								
CH_2-4^I	1.98 m	1.98 m	1.98 m	1.99 m	1.99 m			
CH_2 -4 ^A ,	2.19 s	2.19 s	2.19 s	2.20 s	2.19 s			
CH_2-5^A								
CH_2 -12 ^I	2.56 d							
	$(J_{12, 13} = 7.4 \text{ Hz})$							
2-H ^D	3.09 dd	3.10 dd	_	_	_			
	$(J_{2,3} = 8.1 \text{ Hz})$	$(J_{2,3} = 8.2 \text{ Hz})$						
4-H ^D	3.18 dd	3.20 dd	_	_	_			
	$(J_{4, 5} = 9.5 \text{ Hz})$	$(J_{4, 5} = 9.4 \text{ Hz})$						
5-H ^D	3.29 m	3.29 m	_	_	_			
3-H ^D	3.33 dd	3.34 dd	_	_	_			
	$(J_{3,4} = 9.4 \text{ Hz})$	$(J_{3, 4} = 9.2 \text{ Hz})$						
4-H ^C	3.35-3.43	3.35-3.42	3.35-3.49	3.34-3.49	3.34-3.49			

Table 1—Continued

		Moenomycin						
Assignment	A	A ₁₂	C ₁	C ₃	C ₄			
3-H ^C	3.35-3.43	3.35-3.42	3.35-3.49	3.34-3.49	3.34-3.49			
$2-H^B$	3.35-3.43	3.35-3.42	3.35-3.49	3.34-3.49	3.34-3.49			
5-H ^C	3.43-3.50	3.43-3.51	3.35-3.49	3.34-3.49	3.34-3.49			
2-H ^E	3.43-3.50	3.43–3.51	3.35–3.49	3.34–3.49	3.34–3.49			
4-H ^E	3.43-3.50	3.43–3.51	3.12 dd	3.12 dd	3.34–3.49			
			$(J_{4, 5} = 9.0 \text{ Hz})$ $(J_{3, 4} = 8.4 \text{ Hz})$	$(J_{4, 5} = 8.9 \text{ Hz})$ $(J_{3, 4} = 8.6 \text{ Hz})$				
3-H ^E	3.43-3.50	3.43-3.51	3.35-3.49	3.34 - 3.49	3.34-3.49			
5-H ^E	3.43-3.50	3.43–3.51	3.32 dd	3.27 dd	3.22 m			
J-11	3.43-3.30	3.43-3.31	$(J_{4, 5} = 9.2 \text{ Hz})$	$(J_{4.5} = 9.2 \text{ Hz})$	3.22 m			
			$(J_{5, 6} = 6.1 \text{ Hz})$	$(J_{5, 6} = 6.1 \text{ Hz})$				
2-H ^C	3.50-3.59	3.51-3.59	3.52–3.61	3.53–3.63	3.53-3.63			
3-H ^B	3.50–3.59	3.51–3.59	3.52–3.61	3.53–3.63	3.53–3.63			
6-H _v ^E	3.50–3.59	3.51–3.59	-	-	3.34–3.49			
6-H _v ^D	3.50-3.59		_	_				
4-H ^F	3.30-3.39	3.51–3.59	3.52–3.61	_	_			
	- 2 66 h.d	3.51–3.59		252 262	2 52 2 62			
2-H ^F	3.66 bd $(J \sim 10.5 \text{ Hz})$	3.87–3.97	3.75–3.85	3.53–3.63	3.53–3.63			
6-H _x ^D	3.73 d	3.73 d						
υ-Π _x	$(J_{A, B} = 12.5 \text{ Hz})$	$(J_{A, B} = 10.8 \text{ Hz})$	_	_	_			
1-H _v ^I	*	,	275 295	27/ 29/	277 292			
	3.78–3.85	3.74–3.80	3.75–3.85	3.74–3.84	3.72–3.83			
2-H ^H	3.78–3.85	3.74–3.80	3.75–3.85	3.74–3.84	3.72–3.83			
3-H _y ^H	3.78–3.85	3.74–3.80	3.75–3.85	3.74–3.84	3.72–3.83			
1-H _x ¹	3.89–3.97	3.87–3.97	3.89–3.96	3.90–3.96	3.89–3.94			
3-H _x ^H	3.89–3.97	3.87–3.97	3.89–3.96	3.90–3.96	3.89–3.94			
6-H _x ^E	3.89–3.97	3.87–3.97	-	-	3.89-3.94			
4-H ^B	4.07 bd	4.07 bd	4.07 bd	4.07 bd	4.07 bd			
5-H ^B	$(J_{3,4} = 3.0 \text{ Hz})$ 4.11 d	$(J_{3,4} = 3.0 \text{ Hz})$ 4.11 d	$(J_{3, 4} = 2.8 \text{ Hz})$ 4.11 d	$(J_{3, 4} = 3.0 \text{ Hz})$ 4.11 d	$(J_{3, 4} = 3.1 \text{ Hz})$ 4.11 d			
_	$(J_{4, 5} < 1 \text{ Hz})$	$(J_{4, 5} < 1 \text{ Hz})$						
5-H ^F	4.24 s	4.25 d	4.25 d	4.24 s	4.23 s			
1-H ^D	4.31 d	$(J_{4, 5} = 2.6 \text{ Hz})$ 4.31 d	$(J_{4, 5} = 2.5 \text{ Hz})$	_	_			
2-H ^C	$(J_{1, 2} = 8.0 \text{ Hz})$ 4.38 d	$(J_{1, 2} = 8.0 \text{ Hz})$ 4.40 d	4.41 d	4.42 d	4.34 d			
1-H ^E	$(J_{1, 2} = 8.5 \text{ Hz})$ 4.40 d	$(J_{1, 2} = 8.5 \text{ Hz})$ 4.43 d	$(J_{1, 2} = 8.0 \text{ Hz})$ 4.42 d	$(J_{1, 2} = 8.6 \text{ Hz})$ 4.38 d	$(J_{1, 2} = 8.5 \text{ Hz})$ 4.43 d			
	$(J_{1, 2} = 8.5 \text{ Hz})$ 4.43 d	$(J_{1, 2} = 10.0 \text{ Hz})$	$(J_{1, 2} = 10.5 \text{ Hz})$	$(J_{1, 2} = 8.0 \text{ Hz})$	$(J_{1, 2} \sim 8.5 \text{ Hz})$			
1-H ^B	$(J_{1, 2} = 7.8 \text{ Hz})$	4.43 d $(J_{1, 2} \sim 8.5 \text{ Hz})$	4.44 d $(J_{1, 2} = 7.9 \text{ Hz})$	4.44 d $(J_{1, 2} = 7.9 \text{ Hz})$	4.43 d $(J_{1, 2} = 7.9 \text{ Hz})$			
$CH_2 = 22^I$	4.57 d $(J_{gem} < 1 \text{ Hz})$	$4.57 d$ $(J_{gem} < 1 Hz)$	$4.57 d$ $(J_{gem} < 1 Hz)$	$4.57 d$ $(J_{gem} < 1 Hz)$	4.57 d $(J_{gem} < 1 \text{ Hz})$			
3-H ^F	4.83 d $(J_{2, 3} = 10.4 \text{ Hz})$	4.81 dd $(J_{2,3} = 10.5 \text{ Hz})$	4.80 dd $(J_{2, 3} = 10.4 \text{ Hz})$	4.83 d $(J_{2,3} = 10.4 \text{ Hz})$	4.83 d $(J_{2, 3} = 10.4 \text{ Hz})$			
	(- 4, 5	$(J_{3,4} = 3.1 \text{ Hz})$	$(J_{3, 4} = 3.0 \text{ Hz})$	(-2, 5 222)	(- 2, 5 222)			
17-H ^I	4.97 t	4.98 t	4.98 t	4.98 t	4.98 t			
13-H ^I	$(J_{16, 17} = 6.9 \text{ Hz})$ 5.03 t	$(J_{16, 17} = 6.9 \text{ Hz})$ 5.03 t	$(J_{16, 17} = 7.0 \text{ Hz})$ 5.03 t	$(J_{16, 17} = 6.6 \text{ Hz})$ 5.03 t	$(J_{16, 17} = 6.8 \text{ Hz})$ 5.03 t			
6-H ^I	$(J_{12, 13} = 7.4 \text{ Hz})$ 5.13 dt	$(J_{12, 13} = 7.4 \text{ Hz})$ 5.13 dt	$(J_{12, 13} = 7.4 \text{ Hz})$ 5.13 dt	$(J_{12, 13} = 7.4 \text{ Hz})$ 5.14 dt	$(J_{12, 13} = 7.4 \text{ Hz})$ 5.13 dt			
	$(J_{6, 7} = 15.8 \text{ Hz})$ $(J_{5, 6} = 6.5 \text{ Hz})$	$(J_{6, 7} = 15.7 \text{ Hz})$ $(J_{5, 6} = 6.5 \text{ Hz})$	$(J_{6, 7} = 15.7 \text{ Hz})$ $(J_{5, 6} = 6.7 \text{ Hz})$	$(J_{6, 7} = 15.7 \text{ Hz})$ $(J_{5, 6} = 6.7 \text{ Hz})$	$(J_{6, 7} = 15.8 \text{ Hz})$ $(J_{5, 6} = 6.6 \text{ Hz})$			
2-H ^I	5.20 dd (2 $Js \approx 7.0 \text{ Hz}$)	5.20 dd (2 $Js \approx 6.8 \text{ Hz}$)	5.19 dd (2 $Js \approx 6.8 \text{ Hz}$)	5.20 dd (2 $Js \approx 7.1 \text{ Hz}$)	5.19 dd (2 $Js \approx 7.0$ Hz)			
7-H ^I	5.24 d	5.24 d	5.24 d	5.24 d	5.24 d			
1-H ^F	$(J_{6, 7} = 15.8 \text{ Hz})$ 5.57 dd	$(J_{6, 7} = 15.7 \text{ Hz})$ 5.62 dd	$(J_{6, 7} = 15.7 \text{ Hz})$ 5.58 dd	$(J_{6, 7} = 15.7 \text{ Hz})$ 5.54 dd	$(J_{6, 7} = 15.8 \text{ Hz})$ 5.65 dd			
	$({}^{3}J_{1, P} = 6.0 \text{ Hz})$ $(J_{1, 2} = 3.5 \text{ Hz})$	$({}^{3}J_{1, P} = 6.1 \text{ Hz})$ $(J_{1, 2} = 3.6 \text{ Hz})$	$({}^{3}J_{1, P} = 6.5 \text{ Hz})$ $(J_{1, 2} = 3.7 \text{ Hz})$	$({}^{3}J_{1, P} = 6.1 \text{ Hz})$ $(J_{1, 2} = 3.6 \text{ Hz})$	$(^{3}J_{1, P} = 7.0 \text{ Hz})$ $(J_{1, 2} = 3.6 \text{ Hz})$			

620 HENNIG ET AL.

software) and a Bruker DMX 600 spectrometer (600.13 MHz, processed on an SGI O2 workstation using the X-WINNMR program). Samples consisted of moenomycin A (lyophilized three times from D₂O solution) in D₂O, degassed. For the 600 MHz spectrum the following conditions were used: sample temperature 285 K; 1024 scans without spinning; relaxation delay 2 s; acquisition time 5.5 s; four steady state pulses; 64 K data points, no zero-filling; chemical shifts were calibrated with respect to internal HOD (assuming a chemical shift of 4.75 ppm at 285 K); very slight decouple presaturation of the remaining HOD peak.

Acknowledgements

We thank Professor H. Kessler for his kind support and Renate Herold for skilful technical assistance. Financial support by the Deutsche Forschungsgemeinschaft (Innovationskolleg 'Chemisches Signal und Biologische Antwort'), the Fonds der Chemischen Industrie and Hoechst Marion Roussel (Romainville and Frankfurt) is gratefully acknowledged.

REFERENCES

- For leading references, see (a) O. Ritzeler, L. Hennig, M. Findeisen, P. Welzel and D. Müller, *Tetrahedron* 53, 1665 (1997); (b) O. Ritzeler, L. Hennig, M. Findeisen, P. Welzel, D. Müller, A. Markus, G. Lemoine, M. Lampilas and J. van Heijenoort, *Tetrahedron* 53, 1675 (1997).
- 2. S. C. Stinson, Chem. Eng. News 74(39), 75 (1996).
- Moenomycin A: H.-W. Fehlhaber, M. Girg, G. Seibert, K. Hobert, P. Welzel, Y. van Heijenoort and J. van Heijenoort, Tetrahedron 46, 1557 (1990). Moenomycin C₃ and C₄: J. Scherkenbeck, A. Hiltmann, K. Hobert, W. Bankova, T. Siegels, M. Kaiser, D. Müller, H. J. Veith, H.-W. Fehlhaber, G. Seibert, A. Markus, M. Limbert, G. Huber, D. Böttger, A. Stärk, S. Takahashi, Y. van Heijenoort, J. van Heijenoort and P. Welzel, Tetrahedron 49, 3091 (1993). Moenomycin C₁: M. Hessler-Klintz, K. Hobert, A. Biallass, T. Siegels, M. Hiegemann, A. Maulshagen, P. Welzel, G. Huber, D. Böttger, A. Markus, G. Seibert, A. Stärk, H.-W. Fehlhaber, Y. van Heijenoort and J. van Heijenoort, Tetrahedron 49, 7667 (1993). Moenomycin A₁₂: A. Donnerstag, S. Marzian, D. Müller, P. Welzel, D. Böttger, A Stärk, H.-W. Fehlhaber, A. Markus, Y. van Heijenoort and J. van Heijenoort, Tetrahedron 51, 1931 (1995).
- F. Volke, R. Waschipky, A. Pampel, A. Donnerstag, G. Lantzsch, H. Pfeiffer, W. Richter, G. Klose and P. Welzel, *Chem. Phys. Lipids* 85, 115 (1997).
- A. Donnerstag, L. Hennig, M. Findeisen, P. Welzel and R. Haessner, Magn. Reson. Chem. 34, 1031 (1996).
- 6. D. Lenoir, R. Tschesche, W. Wucherpfennig, G. Huber and H. L. Weidenmüller, *Antimicrob. Agents Chemother.* 144 (1969).
- G. Lantzsch, H. Binder, H. Heerklotz, P. Welzel and G. Klose, Laugmuir, in press.